

Laurent F. Bornaghi,<sup>a</sup> Nicholas B. Drinnan,<sup>b</sup> Tracie Ramsdale,<sup>b</sup> Peter C. Healy<sup>a</sup> and Alan R. White<sup>a\*</sup>

<sup>a</sup>Eskitis Institute of Cell and Molecular Therapy, Griffith University, Nathan, Brisbane 4111, Australia, and <sup>b</sup>Alchemia Pty Ltd, PO Box 6242, Upper Mt Gravatt, Queensland 4122, Australia

Correspondence e-mail: alan.white@griffith.edu.au

#### Key indicators

Single-crystal X-ray study  
 $T = 295$  K  
 Mean  $\sigma(\text{C}-\text{C}) = 0.009$  Å  
 $R$  factor = 0.047  
 $wR$  factor = 0.142  
 Data-to-parameter ratio = 11.2

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

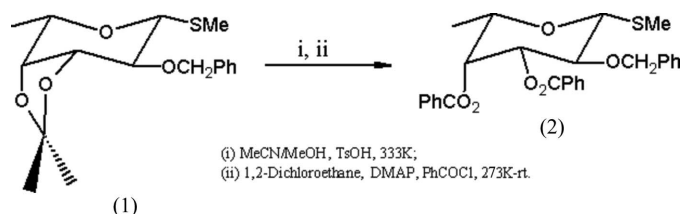
## Methyl 3,4-*O*-dibenzoyl-2-*O*-benzyl-1-thio- $\beta$ -L-fucopyranoside

The title compound,  $\text{C}_{28}\text{H}_{28}\text{O}_6\text{S}$ , is a protected fucose donor, synthesized as part of our investigations into the preparation of complex oligosaccharides. The benzoyl groups are equatorial and axial, while the thioether and ether groups are equatorial. The carboxylate O atoms form a number of intra- and intermolecular  $\text{C}-\text{H}\cdots\text{O}$  interactions.

Received 15 July 2005  
 Accepted 5 August 2005  
 Online 12 August 2005

#### Comment

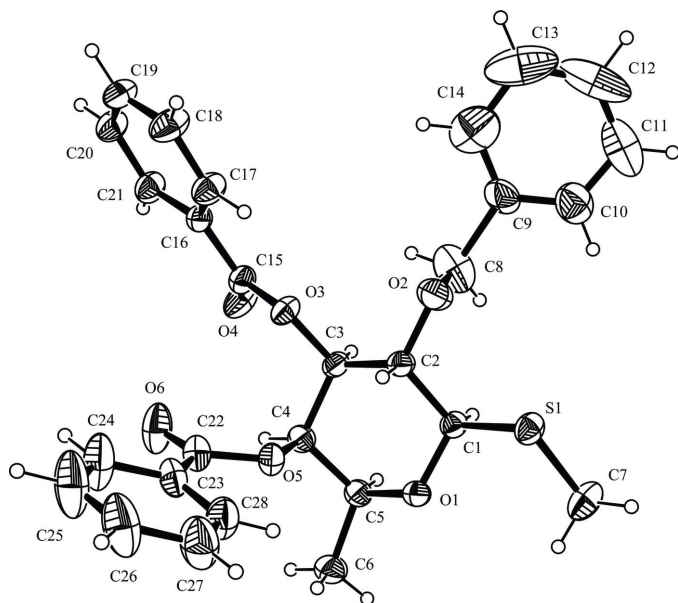
As part of our ongoing investigations into the development of solid-phase syntheses of oligosaccharides, a range of appropriately derivatized donor sugars has been synthesized. It is well established that for stereospecific glycosidic linkage formation, both selective protection of the various hydroxyl groups and a suitably reactive anomeric leaving group are required (Jain & Matta, 1990). When employing fucose donor sugars, it is important to moderate the reactivity of the incumbent fucosyl saccharide, due to the high acid sensitivity of fucosyl glycosidic linkages. It was thought that the title compound, (2), would display reasonable acid stability and also provide excellent anomeric selectivity.



Determination of the crystal structure of (2) showed it to crystallize in the space group  $P2_12_12_1$  with one discrete molecule in the asymmetric unit (Fig. 1). The benzoyl phenyl rings on C3 and C4 are equatorial and axial, respectively. The thioether and ether groups on C1 and C2 are both equatorial. The carboxylate O atoms (O3/O4/O5/O6) form a number of intra- and intermolecular  $\text{C}-\text{H}\cdots\text{O}$  interactions (Table 2).

#### Experimental

Tosic acid, TsOH (100 mg), was added to a solution of methyl 2-*O*-benzyl-3,4-*O*-isopropylidene-1-thio- $\beta$ -L-fucopyranoside, (1) (11.74 g, 0.05 mol) in MeCN/MeOH (1:1, 350 ml). The reaction mixture was then stirred at 333 K overnight, after which time it was cooled to room temperature and neutralized by the addition of triethylamine (2 ml). The solvent was removed *in vacuo*, and the resulting residue was passed through a plug of silica (eluent: ethylacetate/petroleum ethers, 1:1), to give methyl 2-*O*-benzyl-1-thio- $\beta$ -L-fucopyranoside as a



**Figure 1**  
View of compound (2) with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

clear oil (yield 9.90 g, 96%). This compound (6.40 g, 22.53 mmol) was dissolved in 1,2-dichloroethane (100 ml) followed by the addition of dimethylaminopyridine (DMAP, 6.89 g, 56.34 mmol). The solution was cooled to 273 K and benzoyl chloride (6.55 g, 56.34 mmol) was added dropwise. The reaction mixture was returned to room temperature and stirred for 2 h. At this time, DMAP (1.31 g, 11.27 mmol) was added to drive the reaction to completion. After a further hour, the reaction mixture was diluted with chloroform (100 ml) and washed with 10% citric acid solution (2 × 200 ml), saturated sodium hydrogen carbonate solution (2 × 200 ml) and saturated brine solution (2 × 200 ml). The layers were separated and the organic layer dried over Na<sub>2</sub>SO<sub>4</sub>. The solution was filtered, concentrated and purified by column chromatography (ethyl acetate/petroleum ether, 1:5) to provide the title compound, (2), as colourless crystals (yield 9.95 g, 90%). *R*<sub>F</sub> = 0.7 (ethyl acetate/petroleum ether, 1:1); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.98–7.96 (*m*, 2H, Ar), 7.76–7.74 (*m*, 2H, Ar), 7.56–7.52 (*m*, 1H, Ar), 7.45–7.39 (*m*, 3H, Ar), 7.26–7.20 (*m*, 2H, Ar), 7.14–7.07 (*m*, 5H, Ar), 5.59 (*dd*, 1H, H-4, *J*<sub>4,5</sub> = 0.74 Hz), 5.35 (*dd*, 1H, H-3, *J*<sub>3,4</sub> = 3.42 Hz), 4.77, 4.58 (2 × *d*, 2H, PhCH<sub>2</sub>–), 4.50 (*d*, 1H, H-1, *J*<sub>1,2</sub> = 9.23 Hz), 3.95–3.90 (*m*, 1H, H-5), 3.87 (*dd*, 1H, H-2, *J*<sub>2,3</sub> = 9.63 Hz), 2.27 (*s*, 3H, –SCH<sub>3</sub>), 1.23 (*d*, 3H, –CH<sub>3</sub>, *J*<sub>5,6</sub> = 6.16 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 166.11, 165.78 (C=O), 137.67, 133.61, 133.38 (aromatic C1), 130.10, 129.89, 129.82, 129.79, 128.57, 128.51, 128.08 (aromatic C), 85.79 (C1), 75.91 (C2), 75.69 (PhCH<sub>2</sub>), 75.35 (C3), 73.52 (C5), 72.01 (C4), 16.86 (C6), 13.25 (–SCH<sub>3</sub>); TOF HRMS calculated C<sub>28</sub>H<sub>28</sub>O<sub>6</sub>S 492.1607 found *m/z* (ion, relative intensity): 493.1662 ([*M*+H]<sup>+</sup>, 33%).

**Crystal data**

C<sub>28</sub>H<sub>28</sub>O<sub>6</sub>S  
*M*<sub>r</sub> = 492.57  
 Orthorhombic, *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>  
*a* = 11.725 (2) Å  
*b* = 20.396 (3) Å  
*c* = 11.040 (2) Å  
*V* = 2640.1 (8) Å<sup>3</sup>  
*Z* = 4  
*D*<sub>x</sub> = 1.239 Mg m<sup>−3</sup>

Mo *K*α radiation  
 Cell parameters from 25 reflections  
 $\theta$  = 7.7–10.6°  
 $\mu$  = 0.16 mm<sup>−1</sup>  
*T* = 295 K  
 Block, colourless  
 0.35 × 0.20 × 0.15 mm

**Data collection**

Rigaku AFC-7R diffractometer  
 $\omega$  scans  
 Absorption correction: none  
 3971 measured reflections  
 3548 independent reflections  
 2276 reflections with *I* > 2σ(*I*)  
*R*<sub>int</sub> = 0.011

$\theta_{\text{max}}$  = 27.5°  
*h* = 0 → 15  
*k* = −10 → 26  
*l* = −14 → 14  
 3 standard reflections  
 every 150 reflections  
 intensity decay: 0.2%

**Refinement**

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.047  
*wR*(*F*<sup>2</sup>) = 0.142  
*S* = 1.03  
 3548 reflections  
 317 parameters  
 H-atom parameters not refined  
*w* = 1/[σ<sup>2</sup>(*F*<sub>o</sub><sup>2</sup>) + (0.0607*P*)<sup>2</sup> + 1.0014*P*]  
 where *P* = (*F*<sub>o</sub><sup>2</sup> + 2*F*<sub>c</sub><sup>2</sup>)/3

(Δ/σ)<sub>max</sub> < 0.001  
 Δρ<sub>max</sub> = 0.57 e Å<sup>−3</sup>  
 Δρ<sub>min</sub> = −0.25 e Å<sup>−3</sup>  
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0076 (14)  
 Absolute structure: Flack (1983),  
 139 Friedel pairs  
 Flack parameter: −0.19 (15)

**Table 1**

Selected geometric parameters (Å, °).

S1–C1	1.802 (4)	O3–C3	1.442 (5)
S1–C7	1.784 (5)	O3–C15	1.345 (5)
O1–C1	1.432 (5)	O4–C15	1.211 (5)
O1–C5	1.442 (5)	O5–C4	1.441 (5)
O2–C2	1.403 (5)	O5–C22	1.355 (5)
O2–C8	1.295 (8)	O6–C22	1.218 (6)
C1–S1–C7	101.6 (2)	C3–O3–C15	117.8 (3)
C1–O1–C5	112.1 (3)	C4–O5–C22	119.6 (3)
C2–O2–C8	112.3 (4)		

**Table 2**

Hydrogen-bond geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
C1–H1...O4 <sup>i</sup>	0.96	2.55	3.417 (5)	151
C2–H2...O5	0.94	2.54	2.920 (4)	104
C4–H4...O6	0.95	2.36	2.726 (6)	102
C7–H7A...O4 <sup>i</sup>	0.95	2.58	3.452 (6)	152
C17–H17...O3	0.95	2.40	2.738 (5)	100
C28–H28...O5	0.95	2.39	2.721 (6)	100

Symmetry code: (i) +*x* − ½, −*y* + ½, −*z* + 1.

H atoms were positioned geometrically (C–H = 0.90–0.97 Å) and fixed in the refinement, with *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C). The absolute configuration for (2) was assigned assuming unmodified configurations for the chiral centres of (1). The absolute configuration determined with low precision from anomalous scattering effects is in accord with the known configuration of the starting material (Jain & Matta, 1990). The peripheral C atoms in phenyl ring C9–C14 exhibit high anisotropic displacement parameters.

Data collection: *MSC/AFC7 Diffractometer Control Software for Windows* (Molecular Structure Corporation, 1999); cell refinement: *MSC/AFC7 Diffractometer Control Software for Windows*; data reduction: *TEXSAN for Windows* (Molecular Structure Corporation, 2001); program(s) used to solve structure: *TEXSAN for Windows*; program(s) used to refine structure: *TEXSAN for Windows* and *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* (Farrugia, 1997); software used to prepare material for publication: *TEXSAN for Windows* and *PLATON* (Spek, 2003).

The authors acknowledge financial support of this work by Alchemia Pty Ltd, Griffith University and the Eskitis Institute of Cell and Molecular Therapies.

## References

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Jain, R. K. & Matta, K. L. (1990). *Carbohydr. Res.* **208**, 280–286.
- Molecular Structure Corporation (1999). *MSC/AFC7 Diffractometer Control Software for Windows*. Version 1.02. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.
- Molecular Structure Corporation (2001). *TEXSAN for Windows*. Version 1.06. MSC, 9009 New Trails Drive, The Woodlands, TX 77381, USA.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.